

rov [24] on granite must be used with caution since time was not a significant variable in their experiments. Hence, equilibrium was not attained. When time is considered in materials which show increases in  $\sigma$  upon disordering below melting, no significant electrical distinction may be made between a partial melt and a solid of similar composition which is slightly cooler.

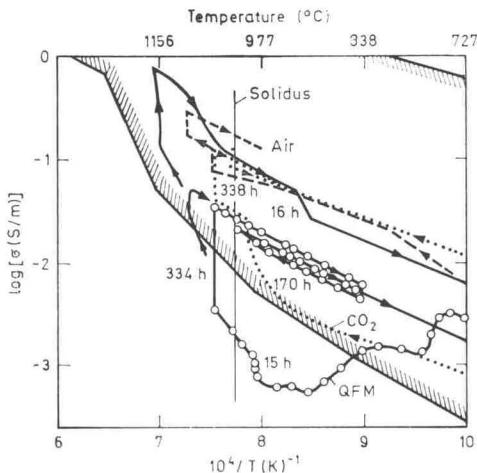


Fig. 7. The electrical conductivity of basalt (DUBA et al. [16]). Lines are coded for  $f_{\text{O}_2}$ : solid lines are for  $\text{CO}_2/\text{CO}$  gas mix near the QFM buffer, dotted lines for pure  $\text{CO}_2$ , and broken lines for air. Constant temperature portions of lines are indicated by heavy shading, with time, in hours, to the side. Normal heating rate is  $100 \text{ }^{\circ}\text{C}/\text{h}$ . The solidus for this material is indicated by the vertical line. The heavy lines with almost vertical shading are the limits of literature  $\sigma$  data for basalt

However, partial melts such as basalt in a solid whose  $\sigma$  is controlled by olivine will show a conductivity contrast of two to four orders of magnitude from the data presented in Figs 3 and 7.

## 6. Conclusion

In summary, temperature and  $f_{\text{O}_2}$  are the most important variables to consider in the interpretation of laboratory  $\sigma$  measurements. In addition, kinetics becomes important if a time-dependent reaction such as order-disorder has a significant effect on  $\sigma$  as in albite and basalt. Pressure is not a significant variable except where it produces crack closure which limits water movement in rocks near the surface [4] and at depth where phase transitions may be involved.

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